

### 3.5 PROCESS SAFETY ASSESSMENT

Process safety is the concern of employers and employees alike. Each company has the obligation to provide its employees with a safe and healthy work environment, while each employee is responsible for his/her own safe personal work habits. An effective process safety program identifies potential workplace hazards and, if possible, seeks to eliminate or at least reduce their potential for harm. In the MHC process of PWB manufacturing, these hazards may be either chemical hazards or process hazards. Chemicals used in the MHC process can be hazardous to worker health and therefore must be handled and stored properly, using appropriate personal protective equipment and safe operating practices. Automated equipment can be hazardous to employees if safe procedures for cleaning, maintaining, and operating are not established and regularly performed. These hazards can result in serious injury and health problems to employees, and potential damage to equipment.

The U.S. Department of Labor and the Occupational Safety and Health Administration (OSHA) have established safety standards and regulations to assist employers in creating a safe working environment and protect workers from potential workplace hazards. In addition, individual states may also have safety standards regulating chemical and physical workplace hazards for many industries. Federal safety standards and regulations affecting the PWB industry can be found in the Code of Federal Regulation (CFR) Title 29, Part 1910 and are available by contacting your local OSHA field office. State and local regulations are available from the appropriate state office. This section of the CTSA presents chemical and process safety concerns associated with the MHC baseline and substitutes, as well as OSHA requirements to mitigate these concerns.

#### 3.5.1 Chemical Safety Concerns

As part of its mission, OSHA's Hazard Communication Standard (29 CFR 1910.1200) requires that chemical containers be labeled properly with chemical name and warning information [.1200(f)], that employees be trained in chemical handling and safety procedures [.1200(h)], and that a MSDS be created and made available to employees for every chemical or formulation used in the workplace [.1200(g)]. Each MSDS must be in English and include information regarding the specific chemical identity of the hazardous chemical(s) involved and the common names. In addition, information must be provided on the physical and chemical characteristics of the hazardous chemical; known acute and chronic health effects and related health information; exposure limits; whether the chemical is a carcinogen; emergency and first-aid procedures; and the identification of the organization preparing the data sheet. Copies of MSDSs for all of the chemicals used must be kept and made available to workers who may come into contact with the process chemicals during their regular work shift.

In order to evaluate the chemical safety concerns of the various MHC processes, MSDSs for 172 chemical products comprising eight MHC technology categories were collected and reviewed for potential hazards to worker safety. The results of that review are summarized and discussed in the categories below. General information on OSHA storage and handling requirements for chemicals in these hazard categories are located in the process safety section of this chapter. For a more detailed description of OSHA storage and handling requirements for

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MHC chemical products in these categories contact your area OSHA field office or state technical assistance program for assistance.

#### **Flammable, Combustible, and Explosive MHC Chemical Products**

A breakdown of MHC chemical products that when in concentrated form are flammable, combustible, explosive, or pose a fire hazard is presented in Table 3.40. The following lists OSHA definitions for chemicals in these categories, and discusses the data presented in the table.

**Table 3.40 Flammable, Combustible, Explosive, and Fire Hazard Possibilities for MHC Processes**

MHC Process	Bath Type	Hazardous Property <sup>a</sup>			
		Flammable	Combustible	Explosive	Fire Hazard
Carbon	Cleaner	2 (2)			
	Conditioner	3 (3)			
	Other (Anti-Tarnish)	2 (2)			
Conductive Ink	Print Ink			5 (5)	
Conductive Polymer <sup>b</sup>	Polymer	1 (3)			
Electroless Copper	Accelerator	1 (5)			
	Anti-Tarnish	2 (4)			
	Cleaner/Conditioner	1 (8)		1 (8)	
	Electroless Copper	2 (25)	1 (25)		1 (25)
	Microetch	1 (9)			
Graphite	Microetch				1 (4)
Non-Formaldehyde Electroless Copper	Accelerator	1 (2)			
	Anti-Tarnish	1 (1)			
	Microetch	1 (4)			
Palladium	Accelerator			1 (10)	1 (10)
	Cleaner/Conditioner	1 (6)	1 (6)		
	Other (Anti-Tarnish)	1 (3)			

<sup>a</sup> Table entries are made in the following format - # of products meeting OSHA definition for the given hazardous property as reported in the products MSDSs (Total # of products in the process bath). A **blank** entry means that none of the products for the specific process bath meet the OSHA reporting criteria for the given property.

Example: For the palladium process accelerator bath, 1 (10) means that one of the ten products in the bath were classified as explosive per OSHA criteria as reported on the products MSDSs.

<sup>b</sup> Hazardous properties based on German equivalent of MSDS, which may not have same reporting requirements of U.S. MSDS.

**Flammable** - A flammable chemical is defined by OSHA [29 CFR 1910.1200(c)] as one of the following:

- An aerosol that, when tested by the method described in 16 CFR 1500.45, yields a flame projection exceeding 18 inches at full valve opening, or a flashback at any degree of valve opening.

- A gas that has: 1) at ambient temperature and pressure, forms a flammable mixture with air at a concentration of 13 percent by volume or less; or 2) when it, at ambient temperature and pressure, forms a range of flammable mixtures with air wider than 12 percent by volume, regardless of the lower limit.
- A liquid that has a flashpoint below 100 °F (37.8 °C), except any mixture having components with flashpoints of 100 °F (37.8 °C) or higher, the total of which make up 99 percent or more of the total volume of the mixture.
- A solid, other than a blasting agent or explosive as defined in 29 CFR 1910.109(a), that is liable to cause fire through friction, absorption of moisture, spontaneous chemical change, or retained heat from manufacturing or processing, or which can be ignited readily and when ignited burns so vigorously and persistently as to create a serious hazard.

Twenty chemical products are reported as flammable according to MSDS data. While all of the products have flashpoints near or below 100 °F, several of the products reported as flammable have flashpoints greater than 200 °F with one as high as 400 °F. Although several chemical products are flammable in their concentrated form, most chemical baths in the MHC process line contain non-flammable aqueous solutions.

**Combustible Liquid** - As defined by OSHA [29 CFR 1910.1200(c)], a liquid that is considered combustible has a flashpoint at or above 100 °F (37.8 °C), but below 200 °F (93.3 °C), except any mixture having components with flashpoints of 200 °F (93.3 °C), or higher, the total volume of which make up 99 percent or more of the total volume of the mixture. Two chemical products have been reported as combustible by their MSDSs, both with flashpoints above 155 °F.

**Explosive** - As defined by OSHA [29 CFR 1910.1200(c)], a chemical is considered explosive if it causes a sudden, almost instantaneous release of pressure, gas, and heat when subjected to sudden shock, pressure, or high temperature. Seven chemical products are reported as explosive by their MSDSs.

**Fire Hazard** - A chemical product that is a potential fire hazard is required by OSHA to be reported on the product's MSDS. According to MSDS data, three chemical products are reported as potential fire hazards.

### **3.5.2 Corrosive, Oxidizer, and Reactive MHC Chemical Products**

A breakdown of MHC chemical baths containing chemical products that are corrosive, oxidizers, or reactive in their concentrated form is presented in Table 3.41. The table also lists process baths that contain chemical products that may cause a sudden release of pressure when opened. The following lists OSHA definitions for chemicals in these categories and discusses the data presented in the table.

**Table 3.41 Corrosive, Oxidizer, Reactive, Unstable, and Sudden Release of Pressure Possibilities for MHC Processes**

MHC Process	Bath Type	Hazardous Property <sup>a</sup>				
		Corrosive	Oxidizer	Reactive	Unstable	Sudden Release of Pressure
Carbon	Cleaner	2 (2)				
	Conditioner	3 (3)				
	Microetch		2 (2)	2 (2)		
Conductive Polymer <sup>b</sup>	Catalyst	2 (3)				
	Conductive Polymer	2 (3)				
	Microetch	1 (1)				
Electroless Copper	Accelerator	1 (5)	1 (5)	3 (5)		
	Catalyst	5 (10)		2 (10)		
	Cleaner/Conditioner	5 (8)		2 (8)		
	Electroless Copper	11 (25)		5 (25)		
	Microetch	3 (9)	5 (9)	2 (9)	1 (9)	1 (9)
	Predip	4 (6)		2 (6)		
Graphite	Fixer	1 (1)				
	Graphite	1 (3)				
	Microetch	2 (4)	1 (4)		1 (4)	
Non-Formaldehyde Electroless Copper	Accelerator		1 (2)	1 (2)		
	Electroless Copper	2 (6)		1 (6)		
	Microetch	2 (4)	2 (4)	2 (4)		1 (4)
Palladium	Accelerator	4 (10)		1 (10)		
	Catalyst	4 (9)		1 (9)		
	Cleaner/Conditioner	1 (6)				
	Microetch			1 (5)	1 (5)	
	Other	2 (3)				
	Predip	1 (4)				

<sup>a</sup> Table entries are made in the following format - # of products meeting OSHA definition for the given hazardous property as reported in the product's MSDSs (Total # of products in the process bath). A **blank** entry means that none of the products for the specific process bath meet the OSHA reporting criteria for the given property. Example: For the graphite process microetch bath, 2 (4) means that two of the four products in the bath were classified as corrosive per OSHA criteria as reported by the products MSDSs.

<sup>b</sup> Hazardous properties based on German equivalent of MSDS, which may not have same reporting requirements of U.S. MSDS.

**Corrosive** - As defined by OSHA (29 CFR 1910.1200 [Appendix A]), a chemical is considered corrosive if it causes visible destruction of, or irreversible alterations in, living tissue by chemical action at the site of contact, as determined by the test method described by the U.S. Department of Transportation 49 CFR Part 173 Appendix A. This term does not apply to chemical action on inanimate surfaces. A review of MSDS data found that 59 MHC chemical products are reported as corrosive in their concentrated form. Some MHC baths may also be corrosive, but MSDSs do not provide data for the process chemical baths once they are prepared.

**Oxidizer** - As defined by OSHA (29 CFR 1910.1200[c]), an oxidizer is a chemical other than a blasting agent or explosive as defined by OSHA [29 CFR 1910.109(a)], that initiates or promotes combustion in other materials, thereby causing fire either of itself or through the release of oxygen or other gases. Twelve chemical products are reported as oxidizers according to MSDS data.

**Reactive** - A chemical is considered reactive if it is readily susceptible to change and the possible release of energy. EPA gives a more precise definition of reactivity for solid wastes. As defined by EPA (40 CFR 261.23), a solid waste is considered reactive if it exhibits any of the following properties: 1) is normally unstable and readily undergoes violent change without detonating; 2) reacts violently or forms potentially explosive mixtures with water; 3) when mixed with water, generates toxic gases, vapors, or fumes in a quantity that can present a danger to human health or the environment (for a cyanide or sulfide bearing waste, this includes exposure to a pH between 2 and 12.5); 4) is capable of detonation or explosive reaction if subjected to a strong initiated source or if heated under confinement; or 5) is readily capable of detonation or explosive decomposition or reaction at standard temperature and pressure. A review of MSDS data found that 25 chemical products from four different MHC processes are considered reactive.

**Unstable** - As defined by OSHA (29 CFR 1910.1200[c]), a chemical is unstable if in the pure state, or as produced or transported, will vigorously polymerize, decompose, condense, or will become self-reactive under conditions of shock, pressure, or temperature. Only three chemical products are reported as unstable according to MSDS data.

**Sudden Release of Pressure** - OSHA requires the reporting of chemical products that, while stored in a container subjected to sudden shock or high temperature, causes a pressure increase within the container that is released upon opening. MSDS data indicated only two chemical products that are potential sudden release of pressure hazards.

### **3.5.3 MHC Chemical Product Health Hazards**

A breakdown of MHC process baths that contain chemical products that are sensitizers, acute or chronic health hazards, or irreversible eye damage hazards in their concentrated form is presented in Table 3.42. Also discussed in this section are MHC chemical products that are potential eye or dermal irritants and suspected carcinogens. The following presents OSHA definitions for chemicals in these categories and discusses the data in Table 3.42 where appropriate.

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**Table 3.42 Sensitizer, Acute and Chronic Health Hazards, and Irreversible Eye Damage Possibilities for MHC Processes**

MHC Process	Bath Type	Hazardous Property <sup>a</sup>			
		Sensitizer	Acute Health Hazard	Chronic Health Hazard	Irreversible Eye Damage
Carbon	Carbon Black		3 (4)	3 (4)	4 (4)
	Cleaner		1 (2)	1 (2)	2 (2)
	Conditioner		3 (3)	3 (3)	2 (3)
	Microetch		2 (2)		2 (2)
	Other (Anti-Tarnish)		2 (2)	2 (2)	2 (2)
Conductive Ink	Print Ink				2 (5)
Conductive Polymer <sup>b</sup>	Catalyst				3 (3)
	Conductive Polymer				2 (3)
	Microetch				1 (1)
Electroless Copper	Accelerator		1 (5)		1 (5)
	Anti-Tarnish		2 (4)	2 (4)	2 (4)
	Catalyst		2 (10)	2 (10)	6 (10)
	Cleaner/Conditioner		1 (8)	1 (8)	3 (8)
	Electroless Copper		5 (25)	4 (25)	13 (25)
	Microetch		3 (9)	1 (9)	4 (9)
	Predip				5 (6)
Graphite	Cleaner/Conditioner		3 (4)	2 (4)	
	Fixer				1 (1)
	Graphite		2 (3)		1 (3)
Non-Formaldehyde Electroless Copper	Microetch		3 (4)	2 (4)	2 (4)
	Accelerator		1 (2)		
	Catalyst		2 (2)	2 (2)	
	Electroless Copper		3 (6)	2 (6)	4 (6)
Organic-Palladium <sup>b</sup>	Microetch		3 (4)	1 (4)	3 (4)
	Conductor				2 (2)
	Postdip				1 (1)
Tin-Palladium	Microetch				1 (1)
	Accelerator		1 (10)		9 (10)
	Catalyst		3 (9)	3 (9)	4 (9)
	Cleaner/Conditioner	2 (6)	1 (6)		2 (6)
	Microetch		2 (5)	2 (5)	3 (5)
	Other		2 (3)		3 (3)
	Acid Dip				1 (1)

<sup>a</sup> Table entries are made in the following format - # of products meeting OSHA definition for the given hazardous property as reported in the product's MSDSs (Total # of products in the process bath). A **blank** entry means that none of the products for the specific process bath meet the OSHA reporting criteria for the given property.

Example: For the palladium process cleaner/conditioner bath, 2 (6) means that two of the six products in the bath were classified as sensitizers per OSHA criteria as reported by the products MSDSs.

<sup>b</sup> Hazardous properties based on German equivalent of MSDS, which may not have same reporting requirements of U.S. MSDS.

**Sensitizer** - A sensitizer is defined by OSHA [29 CFR 1910.1200 Appendix A (mandatory)] as a chemical that causes a substantial proportion of exposed people or animals to develop an allergic reaction in normal tissue after repeated exposure to the chemical. Only two chemical products were reported as sensitizers by MSDS data, both palladium MHC process chemicals.

**Acute and Chronic Health Hazards** - As defined by OSHA (29 CFR 1910.1200 Appendix A), a chemical is considered a health hazard if there is statistically significant evidence based on at least one study conducted in accordance with established scientific principles that acute or chronic health effects may occur in exposed employees. Health hazards are classified using the criteria below:

- Acute health hazards are those whose effects occur rapidly as a result of short-term exposures, and are usually of short duration.
- Chronic health hazards are those whose effects occur as a result of long-term exposure, and are of long duration.

Chemicals that are considered a health hazard include carcinogens, toxic or highly toxic agents, reproductive toxins, irritants, corrosives, sensitizers, hepatotoxins, nephrotoxins, neurotoxins, agents which act on the hematopoietic system, and agents which damage the lungs, skin, eyes, or mucous membranes.

A review of MSDS data found 51 chemical products reported as potentially posing acute health hazards, and 33 chemical products potentially posing chronic health hazards. OSHA does not require reporting of environmental hazards such as aquatic toxicity data, nor are toxicity data on MSDSs as comprehensive as the toxicity data collected for the CTSA. OSHA health hazard data are presented here for reference purposes only, and are not used in the risk characterization component of the CTSA.

**Carcinogen** - As defined by OSHA (29 CFR 1910.1200 Appendix A), a chemical is considered to be a carcinogen if: 1) it has been evaluated by the International Agency for Research on Cancer (IARC), and found to be a carcinogen or potential carcinogen; 2) it is listed as a carcinogen or potential carcinogen in the Annual Report on Carcinogens published by the National Toxicology Program (NTP); or 3) it is regulated by OSHA as a carcinogen. Formaldehyde, which is used as a reducing agent in the electroless copper process, is a suspected human carcinogen. A review of MSDS data found that six chemical products were reported as potential carcinogens. All of the products contain formaldehyde and are utilized in the electroless copper bath of the traditional electroless copper process.

**Dermal or Eye Irritant** - An irritant is defined by OSHA [29 CFR 1910.1200 Appendix A (mandatory)] as a chemical, which is not corrosive, but which causes a reversible inflammatory effect on living tissue by chemical action at the site of contact. A chemical is considered a dermal or eye irritant if it is so determined under the testing procedures detailed in 16 CFR 1500.41- 42. A review of MSDS data found that all but six of the 181 MHC chemical products reviewed are reported as either dermal or eye irritants.

Irreversible Eye Damage - Chemical products that, upon coming in contact with eye tissue, can cause irreversible damage to the eye are required by OSHA to be identified as such on the product's MSDS. A review of MSDS data found that 91 chemical products are reported as having the potential to cause irreversible eye damage.

#### 3.5.4 Other Chemical Hazards

MHC chemical products that have the potential to form hazardous decomposition products are presented below. In addition, chemical product incompatibilities with other chemicals or materials are described, and other chemical hazard categories presented. The following lists OSHA definitions for chemicals in these categories and summarizes the MSDS data where appropriate.

Hazardous Decomposition - A chemical product, under specific conditions, may decompose to form chemicals that are considered hazardous. With few exceptions, the MSDS data for the chemical products in the MHC process indicate the possibility of decomposition to form a potentially hazardous chemical. Each chemical product should be examined to determine its decomposition products so that potentially dangerous reactions and exposures can be avoided. The following are examples of hazardous decomposition of chemical products that are employed in the MHC alternatives:

- When heated, a chemical product used to create an electroless copper bath can generate toxic formaldehyde vapors.
- If allowed to heat to dryness, a graphite bath process chemical could result in gas releases of ammonia, carbon monoxide, and carbon dioxide.
- Thermal decomposition under fire conditions of certain chemical bath constituents of a palladium cleaner/conditioner bath can result in releases of toxic oxide gases of nitrogen and carbon.

Incompatibilities - Chemical products are often incompatible with other chemicals or materials with which they may come into contact. A review of MSDS data found that all of the MHC processes have chemical products with incompatibilities that can pose a threat to worker safety if the proper care is not taken to prevent such occurrences. Incompatibilities reported range from specific chemicals or chemical products, such as acids or cyanides, to other materials, such as rubber or textiles, like wood and leather. Chemical incompatibilities that are common to products from all the MHC processes include acids, alkalis, oxidizers, metals, and reducing agents. Incompatibilities were also found to exist between chemical products used on the same process line. Individual chemical products for each process bath should be closely examined to determine specific incompatibilities and care should be taken to avoid contact with incompatible chemicals and chemical products, textiles, and storage containers.

The following are examples of chemical incompatibilities that exist for chemical products that are employed in the MHC alternatives:

- An electroless copper bath contains chemical products that, when contacted with hydrochloric acid which is present in other electroless copper process baths, will result in reaction forming bis-chloromethyl ether, an OSHA-regulated carcinogen.



- Violent reactions can result when a chemical product of the conductive polymer catalyst bath comes into contact with concentrated acids or reducing agents, both of which are used in PWB manufacturing processes.
- A microetch bath of a graphite process contains chemicals that will react to form hazardous gases when contacted with other chemical products containing cyanides, sulfides, or carbides.
- Hazardous polymerization of a particular conductive ink product can occur when the product is mixed with chemicals products containing amines, anhydrides, mercaptans, or imidazoles.

Other Chemical Hazard Categories - OSHA requires the reporting of several other hazard categories on the MSDSs for chemicals or chemical products that have not already been discussed above. These additional categories include chemical products that are:

- Water-reactive (react with water to release a gas that presents a health hazard).
- Pyrophoric (will ignite spontaneously in air at temperatures below 130 °F).
- Stored as a compressed gas.
- Classified as an organic peroxide.
- Chemicals that have the potential for hazardous polymerization.

A review of MSDS data indicated that none of the chemical products are reported as being water-reactive, pyrophoric, a compressed gas, an organic peroxide, or as having the potential for hazardous polymerization.

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Exposure to chemicals is just one of the safety issues that PWB manufacturers may have to deal with during their daily activities. Preventing worker injuries should be a primary concern for employers and employees alike. Work-related injuries may result from faulty equipment, improper use of equipment, bypassing equipment safety features, failure to use personal protective equipment, and physical stresses that may appear gradually as a result of repetitive motions (i.e., ergonomic stresses). Any or all of these types of injuries may occur if proper safeguards or practices are not in place and adhered to. An effective worker safety program includes:

- An employee training program.
- Employee use of personal protective equipment.
- Proper chemical storage and handling.
- Safe equipment operating procedures.

The implementation of an effective worker safety program can have a substantial impact on business, not only in terms of direct worker safety, but also in reduced operating costs as a result of fewer days of absenteeism, reduced accidents and injuries, and lower insurance costs. Maintaining a safe and efficient workplace requires that both employers and employees recognize and understand the importance of worker safety and dedicate themselves to making it happen.

#### **Employee Training**

A critical element of workplace safety is a well-educated workforce. To help achieve this goal, the OSHA Hazard Communication Standard requires that all employees at PWB manufacturing facilities (regardless of the size of the facility) be trained in the use of hazardous chemicals to which they are exposed. A training program should be instituted for workers, especially those operating the MHC process, who may come into contact with, or be exposed to, potentially hazardous chemicals. Training may be conducted by either facility staff or outside parties who are familiar with the PWB manufacturing process and the pertinent safety concerns. The training should be held for each new employee, as well as periodic retraining sessions when necessary (e.g., when a new MHC process is instituted), or on a regular schedule. The training program should explain to the workers the types of chemicals with which they work and the precautions to be used when handling or storing them; when and how personal protection equipment should be worn; and how to operate and maintain equipment properly.

#### **Storing and Using Chemicals Properly**

Because the MHC process requires handling of a variety of chemicals, it is important that workers know and follow the correct procedures for the use and storage of the chemicals. Much of the use, disposal, and storage information about MHC process chemicals may be obtained from the MSDSs provided by the manufacturer or supplier of each chemical or formulation. Safe chemical storage and handling involves keeping chemicals in their proper place, protected from adverse environmental conditions, as well as from other chemicals with which they may react. Examples of supplier recommended storage procedures found on the MSDSs for MHC chemicals are listed below:

- Store chemical containers in a cool, dry place away from direct sunlight and other sources of heat.
- Chemical products should only be stored in their properly sealed original containers and labeled with the generic name of the chemical contents.
- Incompatible chemical products should never be stored together.
- Store flammable liquids separately in a segregated area away from potential ignition sources or in a flammable liquid storage cabinet.

Some products have special storage requirements and precautions listed on their MSDSs (e.g., relieving the internal pressure of the container periodically). Each chemical product should be stored in a manner consistent with the recommendation on the MSDS. In addition, chemical storage facilities must be designed to meet any local, state, and federal requirements that may apply.

Not only must chemicals be stored correctly, but they must also be handled and transported in a manner which protects worker safety. Examples of chemical handling recommendations from suppliers include:

- Wear appropriate protective equipment when handling chemicals.
- While transporting chemicals, do not use open containers.
- Use only spark-proof tools when handling flammable chemicals.

- Transfer chemicals using only approved manual or electrical pumps to prevent spills created from lifting and pouring.

Proper chemical handling procedures should be a part of the training program given to every worker. Workers should also be trained in chemical spill containment procedures and emergency medical treatment procedures in case of chemical exposure to a worker.

### **Use of Personal Protective Equipment**

OSHA has developed several personal protective equipment standards that are applicable to the PWB manufacturing industry. These standards address general safety and certification requirements (29 CFR Part 1910.132), the use of eye and face protection (Part 1910.133), head protection (Part 1910.135), foot protection (Part 1910.136), and hand protection (Part 1910.138). The standards for eye, face, and hand protection are particularly important for the workers operating the MHC process where there is close contact with a variety of chemicals, of which nearly all irritate or otherwise harm the skin and eyes. In order to prevent or minimize exposure to such chemicals, workers should be trained in the proper use of personal safety equipment.

The recommended personal protective equipment for a worker handling chemicals is also indicated on the MSDS. For the majority of MHC chemicals, the appropriate protective equipment indicated by the MSDS includes:

- Goggles to prevent the splashing of chemical into the eyes.
- Chemical aprons or other impervious clothing to prevent splashing of chemicals on clothing.
- Gloves to prevent dermal exposure while operating the process.
- Boots to protect against chemical spills.

Other items less widely suggested include chemically resistant coveralls and hats. In addition to the personal protective equipment listed above, some MSDSs recommended that other safety equipment be readily available. This equipment includes first aid kits, oxygen supplies (SCBA), and fire extinguishers.

Other personal safety considerations are the responsibility of the worker. Workers should be discouraged from eating or keeping food near the MHC process. Because automated processes contain moving parts, workers should also be prohibited from wearing jewelry or loose clothing, such as ties, that may become caught in the machinery and cause injury to the worker or the machinery itself. In particular, the wearing of rings or necklaces may lead to injury. Workers with long hair that may also be caught in the machinery should be required to securely pull their hair back or wear a hair net.

### **Use of Equipment Safeguards**

In addition to the use of proper personal protection equipment for all workers, OSHA has developed safety standards (29 CFR Part 1910.212) that apply to the actual equipment used in a PWB MHC process. Among the safeguards recommended by OSHA that may be used for conveyORIZED equipment are barrier guards, two-hand trip devices, and electrical safety devices.

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Safeguards for the normal operation of conveyor equipment are included in the standards for mechanical power-transmission apparatus (29 CFR Part 1910.219) and include belts, gears, chains, sprockets, and shafts. PWB manufacturers should be familiar with the safety requirements included in these standards and should contact their local OSHA office or state technical assistance program for assistance in determining how to comply with them.

In addition to normal equipment operation standards, OSHA also has a lockout/tagout standard (29 CFR Part 1910.147). This standard is designed to prevent the accidental start-up of electric machinery during cleaning or maintenance operations that apply to the cleaning of conveyORIZED equipment as well as other operations. OSHA has granted an exemption for minor servicing of machinery provided the equipment has other appropriate safeguards, such as a stop/safe/ready button which overrides all other controls and is under the exclusive control of the worker performing the servicing. Such minor servicing of conveyORIZED equipment can include clearing fluid heads, removing jammed panels, lubricating, removing rollers, minor cleaning, adjusting operations, and adding chemicals. Rigid finger guards should also extend across the rolls, above and below the area to be cleaned. Proper training of workers is required under the standard whether lockout/tagout is employed or not. For further information on the applicability of the OSHA lockout/tagout standard to MHC process operations, contact the local OSHA field office.

#### **Occupational Noise Exposure**

OSHA has also developed standards (29 CFR Part 1910.95) that apply to occupational noise exposure. These standards require protection against the effects of noise exposure when the sound levels exceed certain levels specified in the standard. No data was collected on actual noise levels from MHC process lines, but one PWB manufacturer suggested protective measures may be needed to reduce noise levels from air knife ovens on carbon and graphite lines. This manufacturer installed baffles on his system to reduce noise levels (Kerr, 1997).

## REFERENCES

- ACGIH. 1991. American Conference of Governmental Industrial Hygienists. Documentation of Threshold Limit Values and Biological Exposure Indices, 6th ed. ACGIH, Cincinnati, OH.
- Albright and Wilson. 1992a. Albright and Wilson Americas. 96-Hour LC<sub>50</sub> Bioassay in the Mysid Shrimp, *Mysidopsis bahia*. TSCA sec 8(e) submission 8EHQ-0792-5442 Init.
- Albright and Wilson. 1992b. Albright and Wilson Americas. 48-Hour LC<sub>50</sub> Bioassay in *Daphnia magna*. TSCA sec 8(e) submission 3EHQ-0792-5443 Init.
- AQUIRE. 1995. AQUatic toxicity Information REtrieval database. EPA ERL-Duluth's Aquatic Ecotoxicology Data Systems. U.S. Environmental Protection Agency, Duluth, MN.
- AQUIRE. 1996. AQUatic toxicity Information REtrieval database. Mid-Continent Ecology Division, National Health and Environmental Effects Research Laboratory, Office of Research and Development. U.S. Environmental Protection Agency, Duluth, MN.
- ATSDR. 1990a. Agency for Toxic Substances and Disease Registry. Toxicology Profile for Copper. ATSDR, Atlanta, GA.
- ATSDR. 1990b. Agency for Toxic Substances and Disease Registry. Toxicological Profile for Silver. ATSDR, Atlanta, GA.
- ATSDR. 1992. Agency for Toxic Substances and Disease Registry. Toxicology Profile for Tin. ATSDR, Atlanta, GA.
- ATSDR. 1993. Agency for Toxic Substances and Disease Registry. Technical Report for Ethylene Glycol/Propylene Glycol. Draft for public comment. ATSDR, Atlanta, GA.
- Barnes, D.G. and M. Dourson. 1988. "Reference Dose (RfD): Descriptions and Uses in Health Risk Assessments." *Regulatory Toxicology and Pharmacology*. Vol. 8, p. 471-486.
- Bayes, Martin. 1996. Shipley Company. Personal communication to Jack Geibig, UT Center for Clean Products and Clean Technologies. January.
- Beliles, R.P. 1994. The Metals. In: *Patty's Industrial Hygiene and Toxicology*, 4th ed. G.D. Clayton and F.E. Clayton, Eds. New York: John Wiley & Sons, pp. 2194-2195.
- Berglund, R. and E. Lindh. 1987. "Prediction of the Mist Emission Rate from Plating Baths." *Proc. Am. Electroplaters and Surface Finishers Soc. Ammu. Tech. Conf.*
- Bio/Dynamics Inc. 1984. Inhalation teratology Study in rats and Mice. Final Report. EPA Doc. No. 40-8555049. (As cited in ATSDR 1989 and HSDB 1996).

## REFERENCES

---

- Bureau of Labor Statistics. 1990. Statistical Summary: Tenure with Current Employer as of January 1987. As cited in EPA, 1991b.
- Bureau of Labor Statistics. 1997. Job Tenure Summary. Labor Force Statistics from the current Population Survey. Web site <<http://stats.bls.gov/news.release/tenure.nws.htm>>. Information downloaded on 3/24/97.
- CMA. 1995. Chemical Manufacturers Association. SIDS dossier on isopropanol (IPA). Submitted to the U.S. EPA, Chemical Control Division, Washington, D.C.
- Cohen, L.S., J.M. Friedman, J.W. Jefferson, et al. 1994. A Reevaluation of Risk of In Utero Exposure to Lithium. *J. Am. Med. Assoc.* **271**:146-150.
- Decisioneering, Inc. 1993. Crystal Ball® software.
- Di Margo, David. 1996. Phibro Tech. Personal communication to Lori Kincaid, UT Center for Clean Products and Clean Technologies. June.
- Du Pont. 1995. TSCA sec 8(e) submission 8EHQ-0395-13401. 70% Technical Grade Glycolic Acid. Office of Toxic Substances, U.S. Environmental Protection Agency, Washington, D.C.
- Froiman, Gail. 1996. U.S. Environmental Protection Agency (EPA). Personal communication to Debbie Boger. April.
- Gingell, R. et al. 1994. Glycol ethers and other selected glycol derivatives. In: *Patty's Industrial Hygiene and Toxicology*, 4th ed., Vol. 2A, Toxicology. G.D. Clayton and F.E. Clayton, eds. John Wiley & Sons, New York, pp 2919, 2925-2952.
- Gitlitz, M.H. and M.K. Moran. 1983. Tin Compounds. In: *Kirk-Othmer Encyclopedia of Chemical Technology*, Vol. 23, John Wiley & Sons, New York.
- Graham, B.L., J.A. Dosman, D.J. Cotton, S.R. Weisstock, V.G. Lappi and F. Froh. 1984. Pulmonary function and respiratory systems in potash workers. *J. Occup. Med.* **26**:209-214.
- Greim, H., J. Ahlers, R. Bias, et al. 1994. Toxicity and Ecotoxicity of Sulfonic Acids: Structure-activity Relationship. *Chemosphere* **28**:2203-2236.
- Gross, P. 1985. Biologic activity of hydroxylamine: A review. *CRC Critical Reviews in Toxicology* **14**:87-99.
- Harrington, R.M., R.R. Romano, D. Gates and P. Ridgway. 1995. Subchronic toxicity of sodium chlorite in the rat. *J. Am. Coll. Toxicol.* **14**:21-33.

- Hernandez, O., L. Rhomberg, K. Hogan, C. Siegel-Scott, D. Lai, G. Grindstaff, M. Henry and J.A. Cotruvo. 1994. "Risk assessment for formaldehyde." *Journal of Hazardous Materials*, **39**:161-172.
- HSDB. 1995. Hazardous Substances Data Bank. MEDLARS Online Information Retrieval System, National Library of Medicine. Retrieved 2/20/95.
- HSDB. 1996. Hazardous Substances Data Bank. MEDLARS Online Information Retrieval System, National Library of Medicine.
- IARC. 1984. International Agency for Research on Cancer. Carbon blacks. In: *IARC Monographs on the Evaluation of Carcinogenic Risk of Chemicals to Man*, Vol 33. IARC, Lyon France.
- IARC. 1985. International Agency for Research on Cancer. Hydrogen Peroxide. In: *IARC Monographs on the Evaluation of Carcinogenic Risk of Chemicals to Humans: Allyl Compounds, Aldehydes, Epoxides and Peroxides*, Vol. 36. IARC, Lyon, France.
- IARC. 1987. International Agency for Research on Cancer. *IARC Monographs on the Evaluation of Carcinogenic Risks to Humans*. Overall Evaluations on Carcinogenicity: An Updating of IARC Monographs Volumes 1-43, Supplement 7, p. 71. IARC, Lyon, France.
- IARC. 1989. International Agency for Research on Cancer. *IARC Monographs on the Evaluation of Carcinogenic Risks of Chemicals to Humans. Some Organic Solvents, Resin Monomers and Related Compounds, Pigments and Occupational Exposures in Paint Manufacture and Painting*, Vol. 47. IARC, Lyon, France.
- IARC. 1995. International Agency for Research on Cancer. Wood Dust and Formaldehyde. In: *IARC Monographs on the Evaluation of Carcinogenic Risk of Chemicals to Humans*, Vol. 62. IARC, Lyon, France.
- IARC. 1996. International Agency for Research on Cancer. *IARC Monographs on the Evaluation of Carcinogenic Risks to Humans. Printing Processes, Printing Inks, Carbon Black, and Some Nitro Compounds*, Vol. 65. IARC, Lyon, France.
- Informatics, Inc. 1974. Scientific Literature Reviews on Generally Recognized as Safe (GRAS) Food ingredients: Tartarates. National Technical Information Service.
- Katz, G.V. and D. Guest. 1994. Aliphatic carboxylic acids. In: *Patty's Industrial Hygiene and Toxicology*, 4th ed., Vol. II, Part E, Toxicology. G.D. Clayton and F.E. Clayton, Eds. New York: John Wiley & Sons, pp. 3523-3532.
- Kerr, Michael. 1997. Circuit Center, Inc. Personal Communication to Lori Kincaid, UT Center for Clean Products and Clean Technologies. April 8.

## REFERENCES

---

- Kirwin, C.J., and J.B. Galvin. 1993. Ethers. In: *Patty's Industrial Hygiene and Toxicology*, Vol. 2, Part A, 4th ed. G.D. Clayton and F.E. Clayton, Eds. New York: John Wiley & Sons.
- Lee, S.H. and H. Aleyassine. 1970. Hydrazine toxicity in pregnant rats. *Arch. Environ. Health* **21**:615-619.
- Lington, A.W and C. Bevan. 1994. Alcohols. In: *Patty's Industrial Hygiene and Toxicology*, 4th ed., Vol. 2. G.D. Clayton and F.E. Clayton, Eds. New York: John Wiley & Sons, pp. 2597-2609.
- Mankes, R.F. 1986. Studies on the embryopathic effects of ethanolamine in Long-Evans rats: Preferential Embryopathy in pups contiguous with themale siblings in utero. *Teratogen Carcinogen Mutagen* **6**: 403-417.
- Nelson, K.W., J.F. Ege, Jr., M. Ross, et al. 1943. Sensory response to certain industrial solvent vapors. *J. Ind. Hyg. Toxicol.* **25**:282-285. (Cited in ACGIH 1991).
- NIOSH. 1976. National Institute for Occupational Safety and Health. *A Guide to Industrial Respiratory Protection*. Cincinnati, OH: NIOSH, U.S. Department of Health and Human Services. NEW Pub. 76-189.
- NIOSH. 1994. National Institute for Occupational Safety and Health. *NIOSH Pocket Guide to Chemical Hazards*.
- NTP. 1992. National Toxicology Program. *Toxicology and Carcinogenesis Studies of Resorcinol*. Tech Rep Ser. No. 403.
- NTP. 1994. National Toxicology Program. NTP Technical Report on the Toxicology and Carcinogenesis Studies of Triethanolamine (CAS No. 102-71-6) in F344/N Rats and B6C3F<sub>1</sub> Mice (Dermal Studies).
- Opresko, D.M. 1995. Toxicity Summary for Lithium. Prepared for Oak Ridge Reservation Environmental Restoration Program, Oak Ridge National Laboratory, Oak Ridge, TN.
- Pendergrass, J.A. 1983. Graphite. In: *Encyclopedia of Occupational Health and Safety*, 3rd ed., Vol. 1. L. Parmegiani, Ed. International Labour Office, Geneva, pp. 978-979.
- Perry, W.G., F.A. Smith and M.B. Kent. 1994. Ternary salts of alkali metals containing oxygen. In: *Patty's Industrial Hygiene and Toxicology*, 4th ed., Vol. II, Part F. G.D. Clayton and F.E. Clayton, Eds. New York: John Wiley & Sons, pp. 4496-4505.
- Pierce, J.O. 1994. Alkaline metals. In: *Patty's Industrial Hygiene and Toxicology*, 4th ed., Vol. II, Part A. G.D. Clayton and F.E. Clayton, Eds. New York: John Wiley & Sons, pp. 769-773.



- Price, J.C., C.A. Kimmel, J.D. George and M.C. Marr. 1987. The developmental toxicity of diethylene glycol dimethyl ether in mice. *Fund. Appl. Toxicol.* **8**:115-126.
- Reynolds, J.E.F. (Ed.) 1982. Ammonium chloride. In: *Martindale the Extra Pharmacopoeia*, 28th ed. The Pharmaceutical Press, London, pp. 686-689.
- Risk\*Assistant™ 2.0. 1995. Hampshire Research Institute.
- RM1. 1992. "RAB Disposition of Diethylene Glycol Butyl Ether (DGBE) and Diethylene Glycol Butyl Ether Acetate (DGBA) for RM1 Presentation." Agi Revesz, January 29. ECAD/RAB.
- RM2. 1996. RM1 recommendations for isopropanol. L. Anderson, U.S. EPA Chemical Screening and Risk Assessment Division. February 9. Draft.
- Robinson, R.B., C.D. Cox, N.D. Jackson and M.B. Swanson. 1997. "Estimating Worker Inhalation Exposure to Chemicals From Plating Baths at Printed Wiring Board Facilities." Presented at the 1997 Canadian Society of Civil Engineers - American Society of Civil Engineers Environmental Engineering conference. July. Edmonton, Alberta, Canada. July.
- Rohm and Haas. 1992. Rohm and Haas Company. A Teratologic Evaluation of EGMBE in Fischer 344 Rats and New Zealand White Rabbits Following Inhalation Exposure (Report No. 84RN-1104). Cover letter dated August 13, 1992. 8EHQ-92-9728 init.; 88920008030.
- RTECS. 1995. Registry of Toxic Effects of Chemical Substances. MEDLARS Online Information Retrieval System, National Library of Medicine. Retrieved 2/95.
- RTECS. 1996. Registry of Toxic Effects of Chemical Substances. MEDLARS Online Information Retrieval System, National Library of Medicine.
- Schroeder, H.A. and M. Mitchener. 1971. Scandium, chromium (VI), gallium, yttrium, rhodium, palladium, indium in mice: effects on growth and life span. *J. Nutr.* **101**:1431-1438.
- Shell Oil. 1992a. Shell Oil Company. HSE-84-0073 (002333). Final report teratologic evaluation of ethylene glycol (CAS No. 107-21-1) administered to CD-1 mice on gestational days 6 through 15. Cover letter dated March 5, 1992. 8EHQ-0592-3462 Init.
- Shell Oil. 1992b. Shell Oil Company. HSE-84-0114 (002627). Evaluation of the teratogenic potential of ethylene glycol aerosol in the CD rat and the CD-1 mouse. Cover letter dated March 9, 1992. 8EHQ-0492-3366 Init.
- Shepard, T.H. 1986. *Catalog of Teratogenic Agents*, 5th ed. The Johns Hopkins University Press, Baltimore, p. 38.

## REFERENCES

---

- SIDS. 1995. SIDS Dossier on Iosopropanol (IPA) submitted by the Chemical Manufacturers Association to U.S. EPA Chemical Control Division. March 28.
- Stokinger, H.E. 1981. Fluoroboric acid and salts. In: *Patty's Industrial Hygiene and Toxicology*, 3rd rev. ed., Vol. 2B. G.D. Clayton and F.E. Clayton, eds. Wiley Interscience, New York, pp. 2943-2944.
- Szabo, K.T. 1970. Teratogenic Effects of Lithium Carbonate in the Foetal Mouse. *Nature* **225**:73-75.
- Tewe, O.O. and J.H. Maner. 1981. Performance and pathophysiological changes in pregnant pigs fed cassava diet containing different levels of cyanide. *Res. Vet. Sci.* 30: 147-151.
- TOXLINE. 1995. MEDLARS Online Information Retrieval System, National Library of Medicine. Retrieved 2/95.
- TR-Metro. 1994. TR-AMC Chemicals. Acute Toxicity of Sodium Chlorite to the Sheepshead Minnow; Effect of Sodium Chloride on New Shell Growth in Eastern Oyster. TSCA sec 8(e) submission 8EHQ-0894-13008.
- Trochimowicz, H.J., G.L. Kennedy and N.D. Krivanek. 1994. Heterocyclic and miscellaneous nitrogen compounds. In: *Patty's Industrial Hygiene and Toxicology*, 4th ed., Vol. 2. G.D. Clayton and F.E. Clayton, Eds. New York: John Wiley & Sons.
- Tyl, R.W., B. Ballantyne, L.C. Fisher, D.L. Fait, D.E. Dodd, D.R. Klonne, I.M. Pritts and P.E. Losco. 1995. "Evaluation of the Developmental Toxicity of Ethylene Glycol Aerosol in CD-1 Mice by Nose-Only Exposure." Bushy Run Research Center, Export, PA and Union Carbide Corporation, Danbury, CT.
- Union Carbide. 1991. Union Carbide Chemicals and Plastics Company, Inc. Evaluation of the teratogenic potential of ethylene glycol aerosol in the CD rat and the CD-1 mouse. Cover letter dated November 7, 1991. 8EHQ-1191-1508 Init.
- U.S. Air Force. 1990. Copper - Elemental Copper. In: *The Installation Restoration Toxicology Guide*. Wright-Patterson Air Force Base, OH. Vol. 5, pp. 77-1 - 77-44.
- U.S. Borax Co. 1992. Developmental Toxicity Study with Rabbits. Special Review and Reregistration Submission to U.S. EPA. 8EHQ-0192-2002.
- U.S. Environmental Protection Agency (EPA). 1980. *Ambient Water Quality Criteria for Cyanides*. EPA Office of Water and Standards, Criteria and Standards Division, Washington, D.C.
- U.S. Environmental Protection Agency (EPA). 1984a. Health Effects Assessment for Copper. Office of Research and Development, Office of Emergency and Remedial Response, Washington, DC., Office of Health and Environmental Assessment, Environmental Criteria and Assessment Office, Cincinnati, OH.

- U.S. Environmental Protection Agency (EPA). 1984b. Health Effects Assessment for Glycol Ethers. Environmental Criteria and Assessment Office, Cincinnati, OH, 77 pages.
- U.S. Environmental Protection Agency (EPA). 1984c. *Estimating Concern Levels for Concentrations of Chemical Substances in the Environment*. EPA Office of Pollution Prevention and Toxics, Health and Environmental Review Division, Environmental Effects Branch, Washington, D.C.
- U.S. Environmental Protection Agency (EPA). 1985a. Health and Environmental Effects Profile for 2-Ethoxyethanol. EPA Environmental Criteria and Assessment Office, Cincinnati, OH.
- U.S. Environmental Protection Agency (EPA). 1985b. Health and Environmental Effects Profile for Formaldehyde. EPA Office of Solid Waste and Emergency Response, Washington, D.C. ECAO-CIN-P142.
- U.S. Environmental Protection Agency (EPA). 1986. Health and Environmental Effects Profile for N,N-Dimethylformamide. EPA Environmental Criteria and Assessment Office, Cincinnati, OH. ECAO-CIN-P158.
- U.S. Environmental Protection Agency (EPA). 1987. Reportable Quantity for Ethylenediamine Tetraacetic Acid (EDTA). Environmental Criteria and Assessment Office, Office of Health and Environmental Assessment, Cincinnati, OH, 5 pp. ECAO-CIN-R410.
- U.S. Environmental Protection Agency (EPA). 1988. Reportable Quantity Document for Hydrogen Peroxide. EPA Office of Solid Waste and Emergency Response, Environmental Criteria and Assessment Office, Cincinnati, OH, 9 p.
- U.S. Environmental Protection Agency (EPA). 1989. *Risk Assessment Guidance for Superfund, Volume I: Human Health Evaluation Manual (Part A)*.
- U.S. Environmental Protection Agency (EPA). 1990. *Exposure Factors Handbook*. U.S. EPA, Office of Health and Environmental Assessment, Washington, D.C. EPA 600-8-89/043.
- U.S. Environmental Protection Agency (EPA). 1991a. *Chemical Engineering Branch Manual for the Preparation of Engineering Assessments*. EPA Office of Toxic Substances, Washington, D.C.
- U.S. Environmental Protection Agency (EPA). 1991b. Human Health Evaluation Manual, Supplemental Guidance: "Standard Default Exposure Factors." EPA Office of Solid Waste and Emergency Response, Washington D.C.
- U.S. Environmental Protection Agency (EPA). 1991c. Formaldehyde Risk Assessment Update. EPA Office of Toxic Substances, Washington, D.C.

## REFERENCES

---

- U.S. Environmental Protection Agency (EPA). 1992a. *Dermal Exposure Assessment: Principles and Applications, Interim Repot.* EPA Office of Research and Development, Washington, D.C. EPA/600/18-91/011B.
- U.S. Environmental Protection Agency (EPA). 1992b. *Guidelines for Exposure Assessment.* Washington, D.C. EPA 600-2-92-001.
- U.S. Environmental Protection Agency (EPA). 1994a. Health Effects Assessment Summary Tables FY-1994 Annual. HEAST Table 1: Subchronic and Chronic Toxicity (Other than Carcinogenicity). EPA Environmental Criteria and Assessment Office, Cincinnati, OH.
- U.S. Environmental Protection Agency (EPA). 1994b. *ECOSAR: A Computer Program for Estimating the Ecotoxicity of Industrial Chemicals Based on Structure Activity Relationships: User's Guide.* EPA Office of Pollution Prevention and Toxics, Washington, D.C. EPA-784-R-93-002.
- U.S. Environmental Protection Agency (EPA). 1995a. *Printed Wiring Board Pollution Prevention and Control: Analysis of Survey Results.* Design for the Environment Printed Wiring Board Project. EPA Office of Pollution Prevention and Toxics, Washington, D.C. EPA 774-R-95-006. September.
- U.S. Environmental Protection Agency (EPA). 1995b. Chemical Summary for Formaldehyde. EPA Office of Pollution Prevention and Toxics, Washington, D.C.
- U.S. Environmental Protection Agency (EPA). 1995c. Integrated Risk Information System (IRIS) Online. EPA Office of Health and Environmental Assessment, Cincinnati, OH.
- U.S. Environmental Protection Agency (EPA). 1995d. Health Effects Assessment Summary Tables Annual Update, FY-95. EPA Environmental Criteria and Assessment Office, Cincinnati, OH.
- U.S. Environmental Protection Agency (EPA). 1996a. "Proposed Guidelines for Carcinogen Risk Assessment." EPA Office of Research and Development, Washington, D.C. EPA/600/P-92/1003C.
- U.S. Environmental Protection Agency (EPA). 1996b. Integrated Risk Information System (IRIS) Online. EPA Office of Health and Environmental Assessment, Cincinnati, OH.
- U.S. Environmental Protection Agency (EPA). 1996c. Phenol. Integrated Risk Information system (IRIS) Online. Office of Health and Environmental Assessment, Cincinnati, OH. Retrieved online 8/30/96.
- U.S. Environmental Protection Agency (EPA). 1997. Exposure Factors Handbook. EPA Office of Research and Development, Washington, DC. EPA/600/P-951002Fa.
- Verschueren K. (Ed.). 1983. *Handbook of Environmental Data on Organic Chemicals*, 2nd ed. New York: Van Nostrand Reinhold Company.

- Verschueren, K. 1996. *Handbook of Environmental Data on Organic Chemicals*, 3rd ed. New York: Van Nostrand Reinhold Company.
- Wiese, W.H. and B.J. Skipper. 1986. Survey of reproductive outcomes in uranium and potash mine workers: Results of first Analysis. *Annals of the American Conference of Governmental Industrial Hygienists* **19**:187-192. (As cited in TOXLINE, 1995).
- Williamson, Tracy. 1996. U.S. Environmental Protection Agency (EPA). Personal communication to Jack Geibig, UT Center for Clean Products and Clean Technologies. November.
- Woodiwiss, F.S. and G. Fretwell. 1974. The toxicities of sewage effluents, industrial discharges and some chemical substances to brown trout (*Salmo trutta*) in the Trent River Authority area. *Water Pollut. Control* **73**:396-405. (Cited in AQUIRE 1995)
- Wu, Q.Z. 1990. Teratogenic studies on stannous chloride in rats. *Chung-Hua-Yu-Fang-I-Hsueh-Tsa-Chih*. **24**(1): 19-21. January. (As cited in Toxline, 1995.)
- Young, R.A. 1992. Reportable Quantity Document for Sodium Sulfate. Prepared for Environmental Criteria and Assessment Office, U.S. Environmental Protection Agency, Cincinnati, OH.